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# Engineered Particulates for Co-Firing of Diverse Feedstocks

*Joseph J. McCarthy*

mccarthy@engrng.pitt.edu

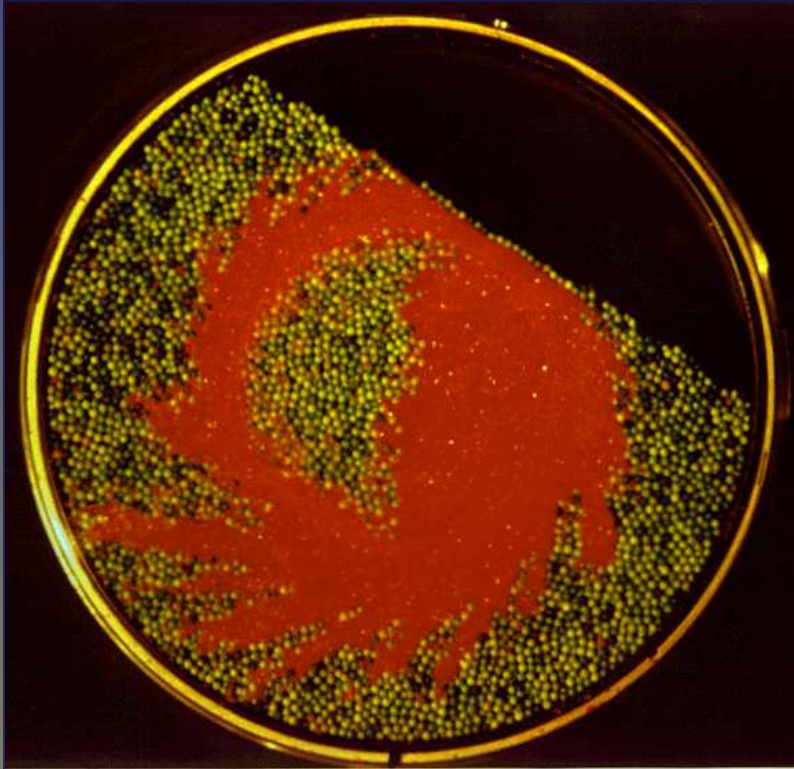
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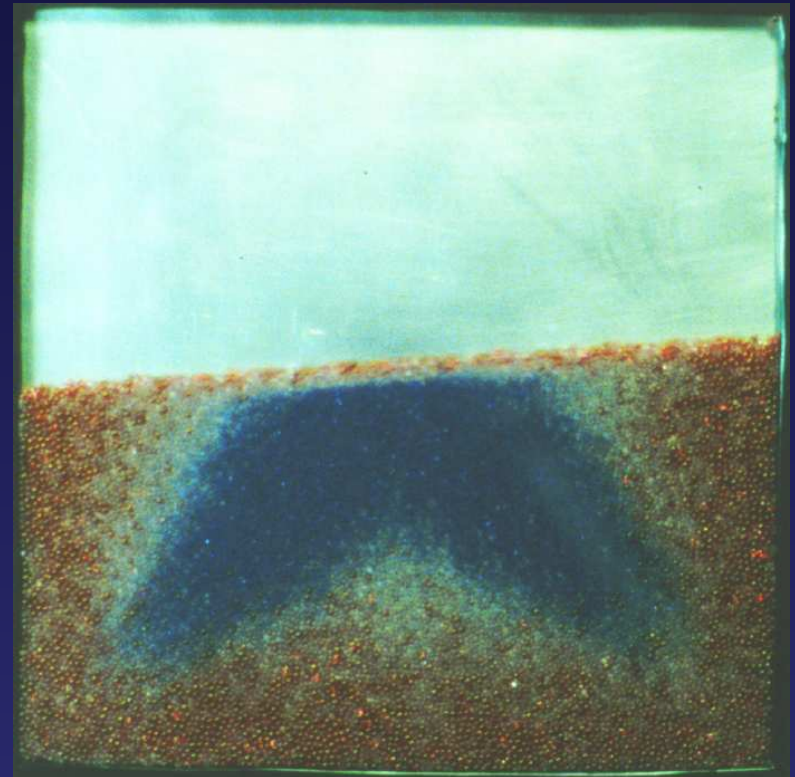


# Motivation



(AICHEJ, 1996)

## Segregation &



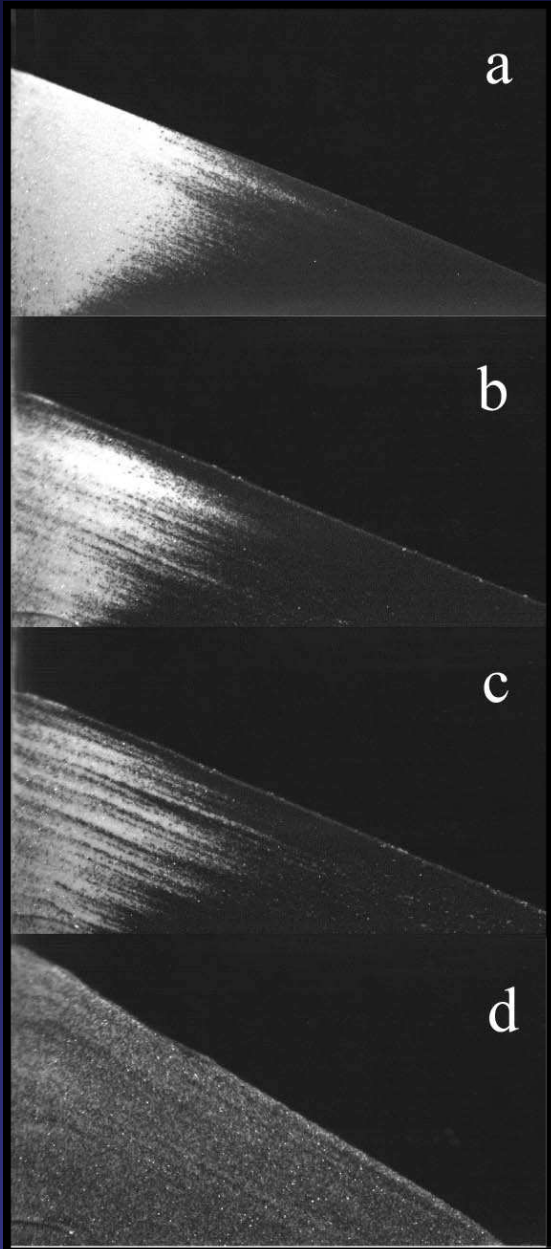
(PNAS, 1996)

## Pattern Formation

- Differences in mechanical properties
- Inconsistent feed (possibly burn quality)



# Motivation (cont.)



Cohesion sometimes segregation!

- Known long – recent quantitative studies
- can this be controlled?



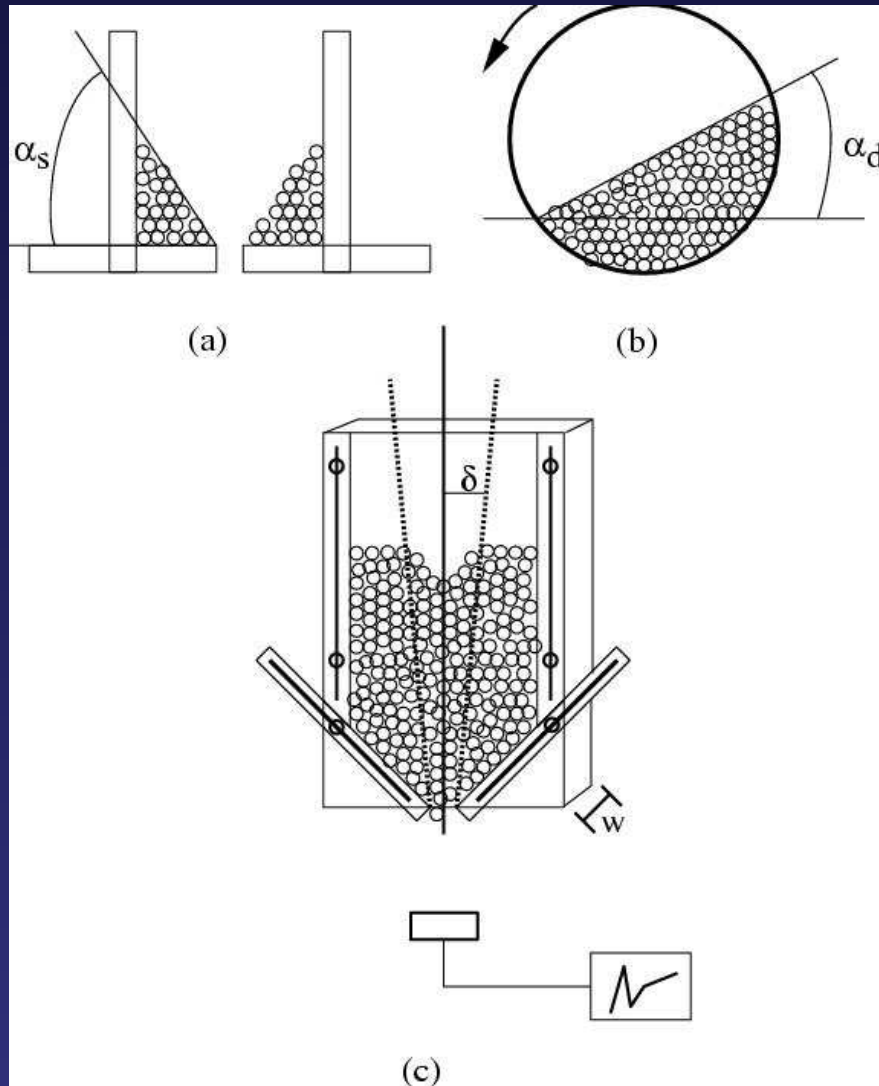
# Outline

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- Cohesive Characterization Tools
  - Static systems
  - Sheared systems
  - Gas-solid systems
- Mixing/Segregation – Aggregate Formation
  - Psuedo-Static systems
  - Sheared systems
  - Gas-solid systems



# Cohesive Characterization Exps.



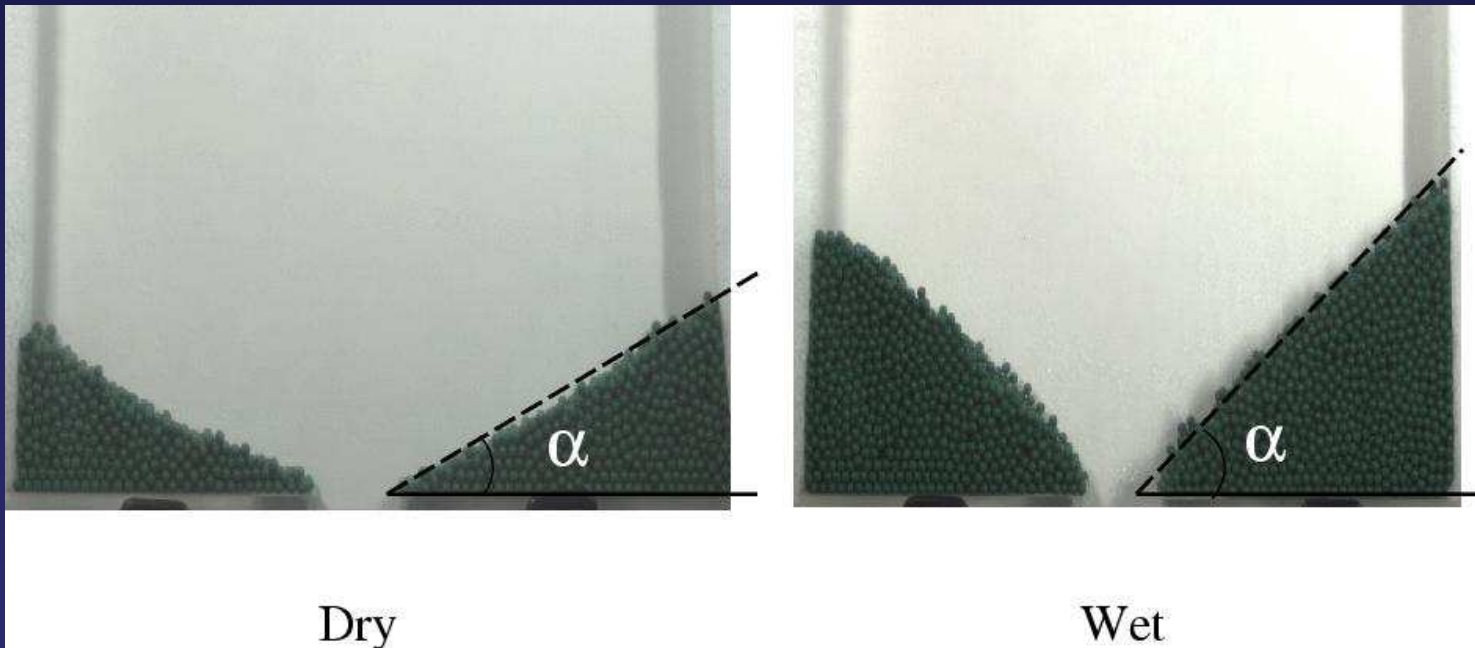
- “Perfectly” static (a)
- Prototypical granular flows (b) and (c)



# Static Systems

## Relevant Forces

- Particle self-weight ( $W$ )
- Capillary Force ( $F_c$ )

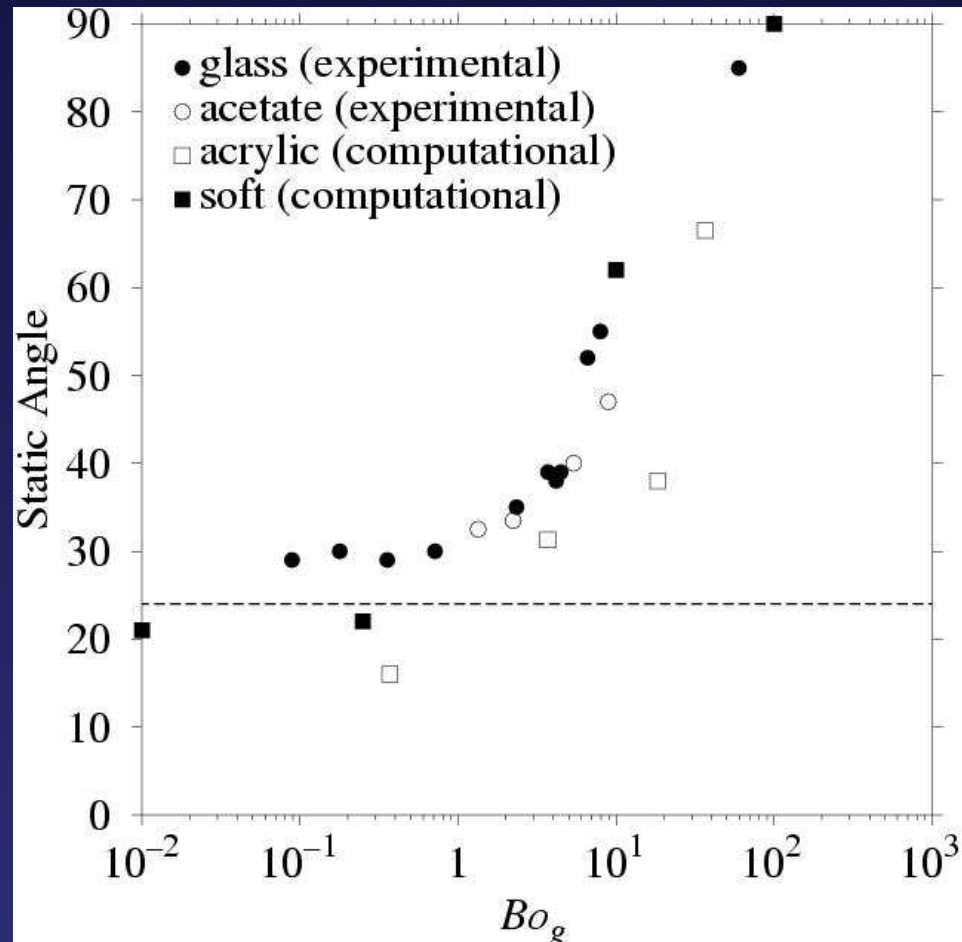
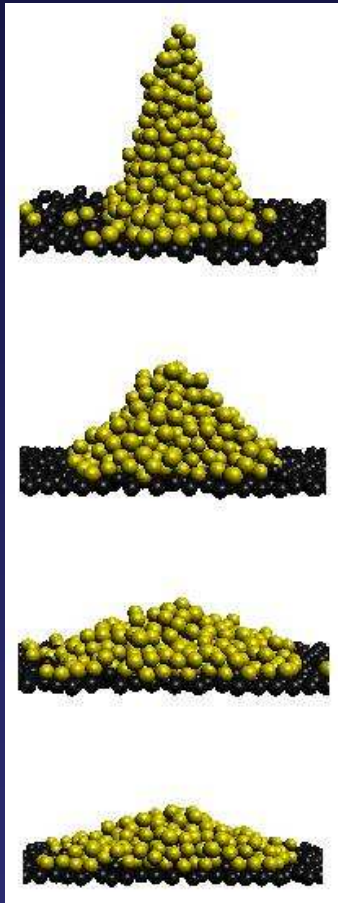


$$Bo_g = \frac{F_c}{W} = \frac{3\gamma}{2R^2\rho g}$$





# Angle of Repose - Static Heap

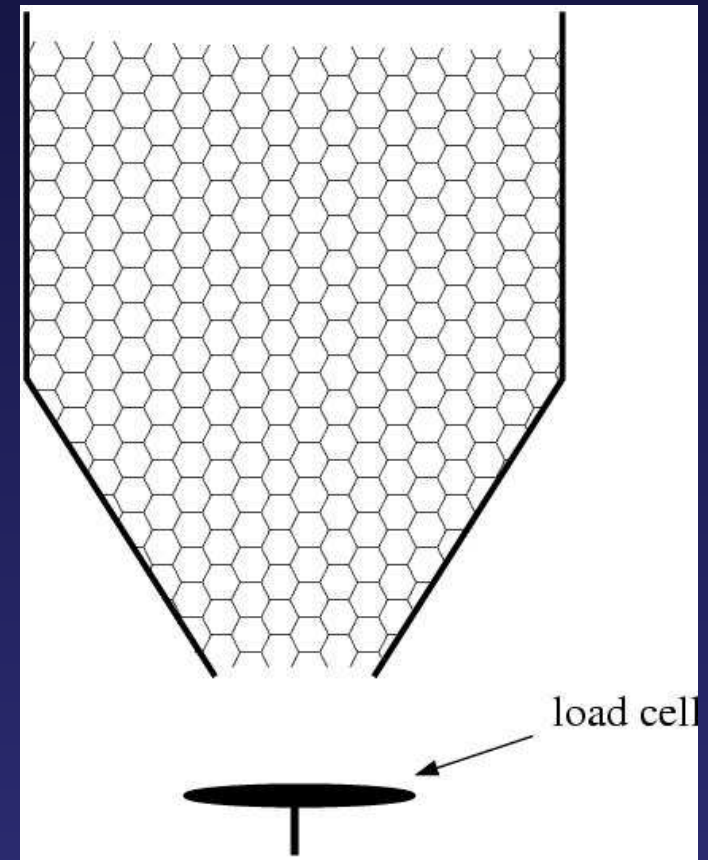
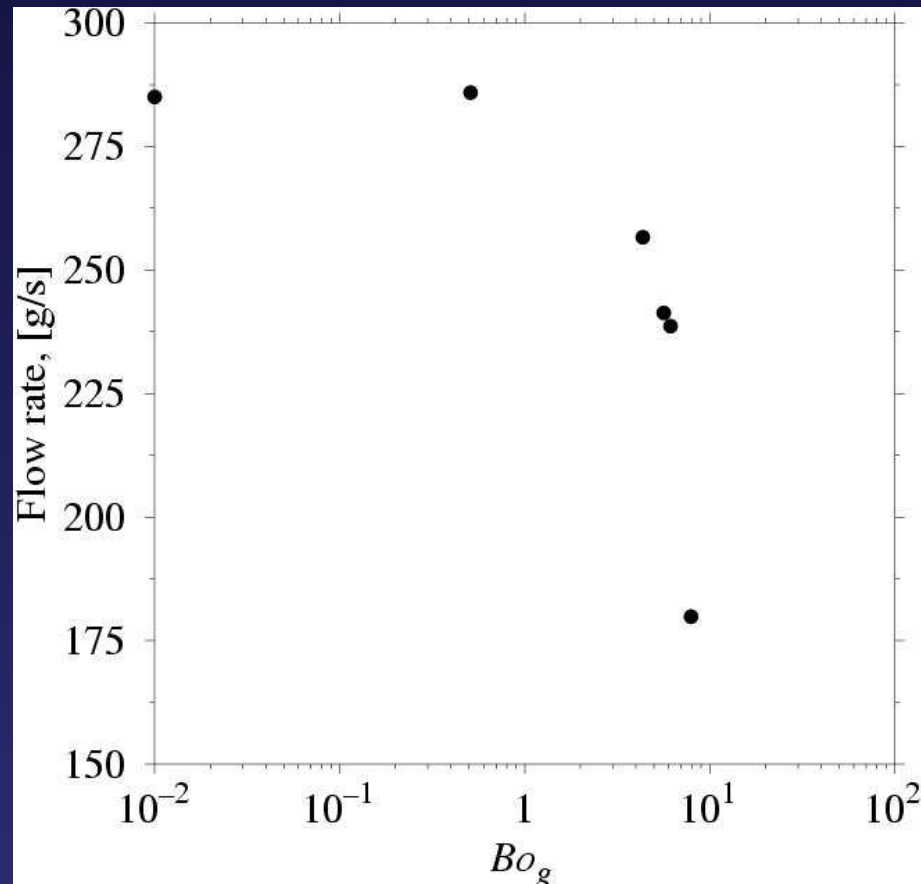


(Pow. Technol., 2000)

- Sharp transition near  $Bo_g = 1$
- Vastly different materials all follow same trend



# Hopper Discharge

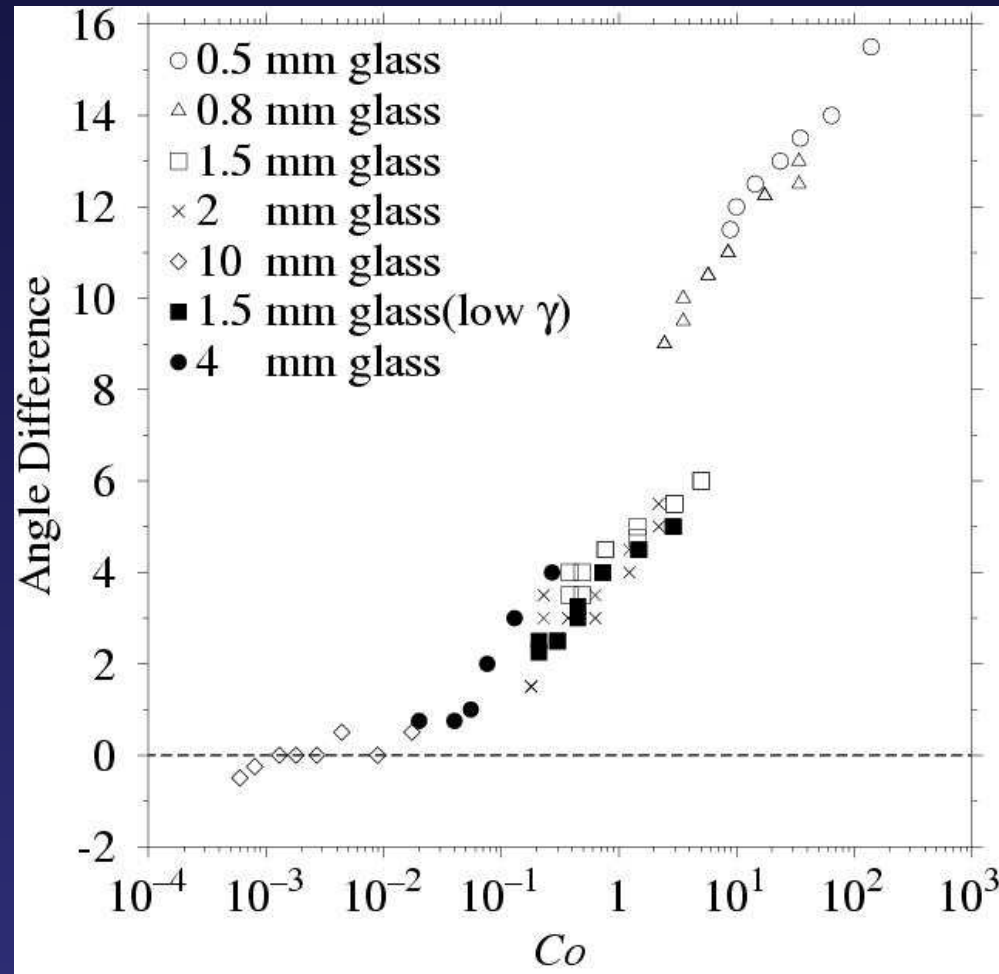


- Sharp transition above  $Bo_g = 1$
- Essentially no flow thereafter (arching occurs)





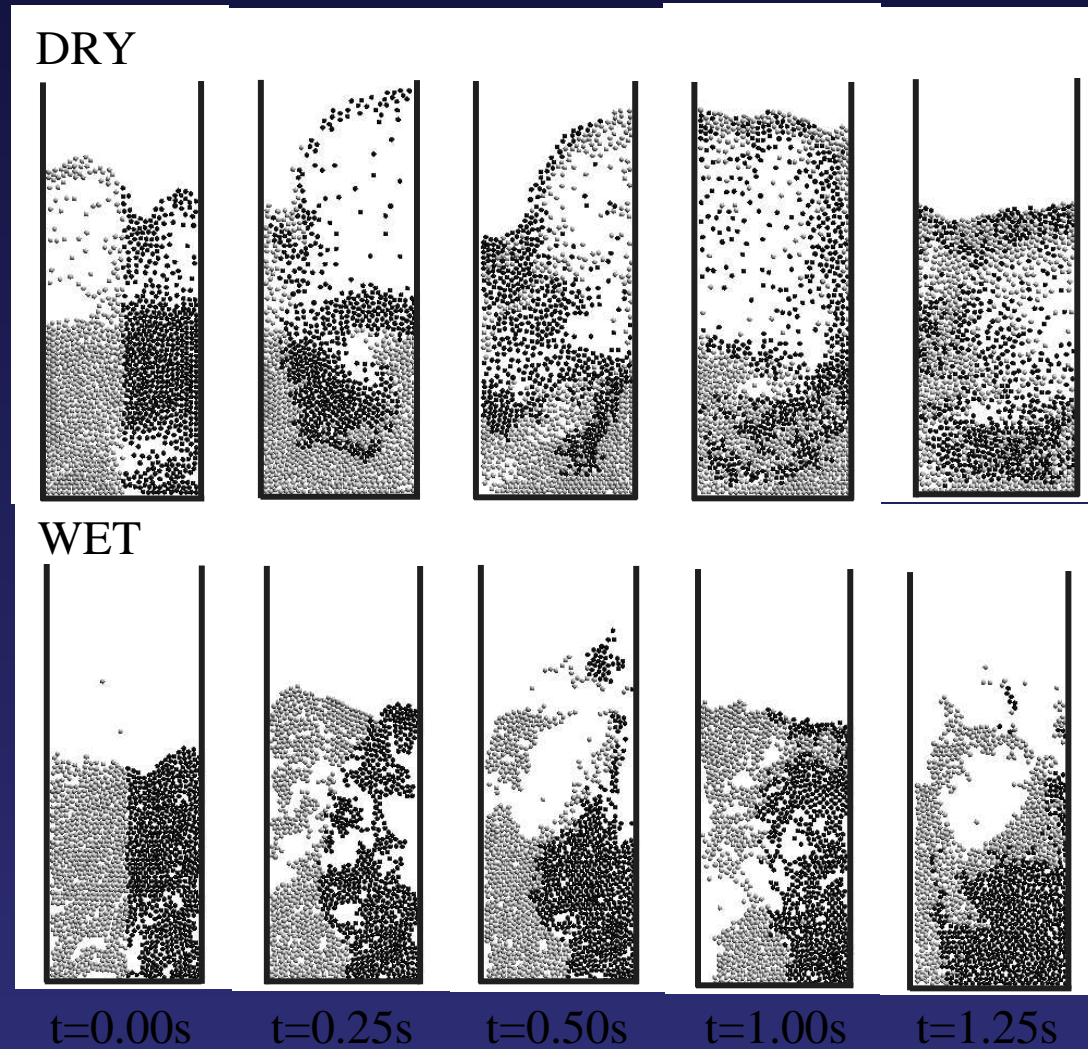
# Collision Number



- Sharp transition near  $Co = 1$



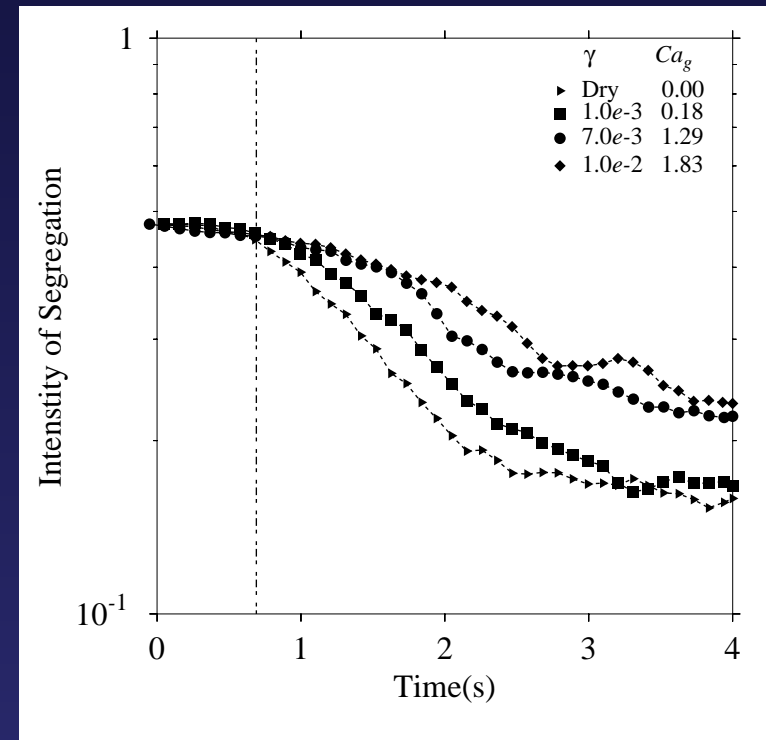
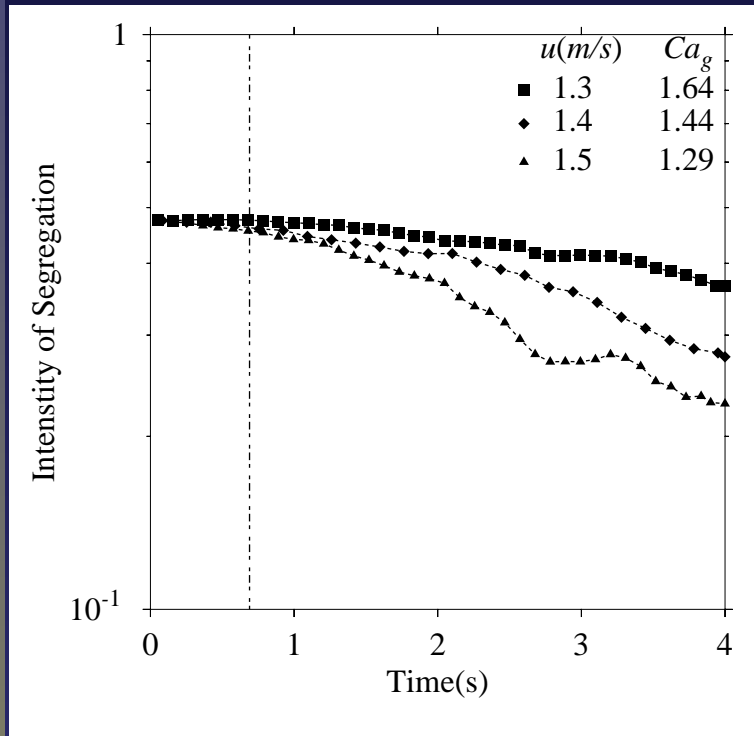
# Gas-solid Flows



- Mixing rate changes with cohesion
- Minimum fluidization velocity *also* changes



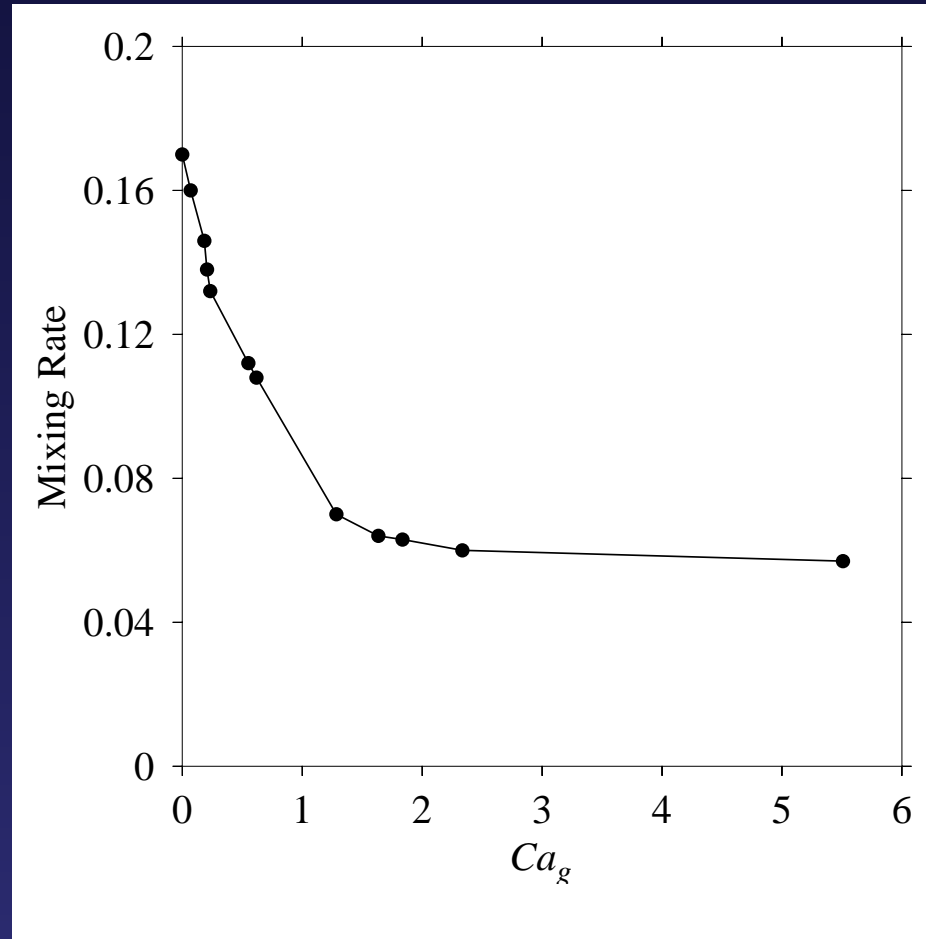
# Gas-solid Mixing



- Increase in fluidization velocity  $\approx$  decrease in  $\gamma$



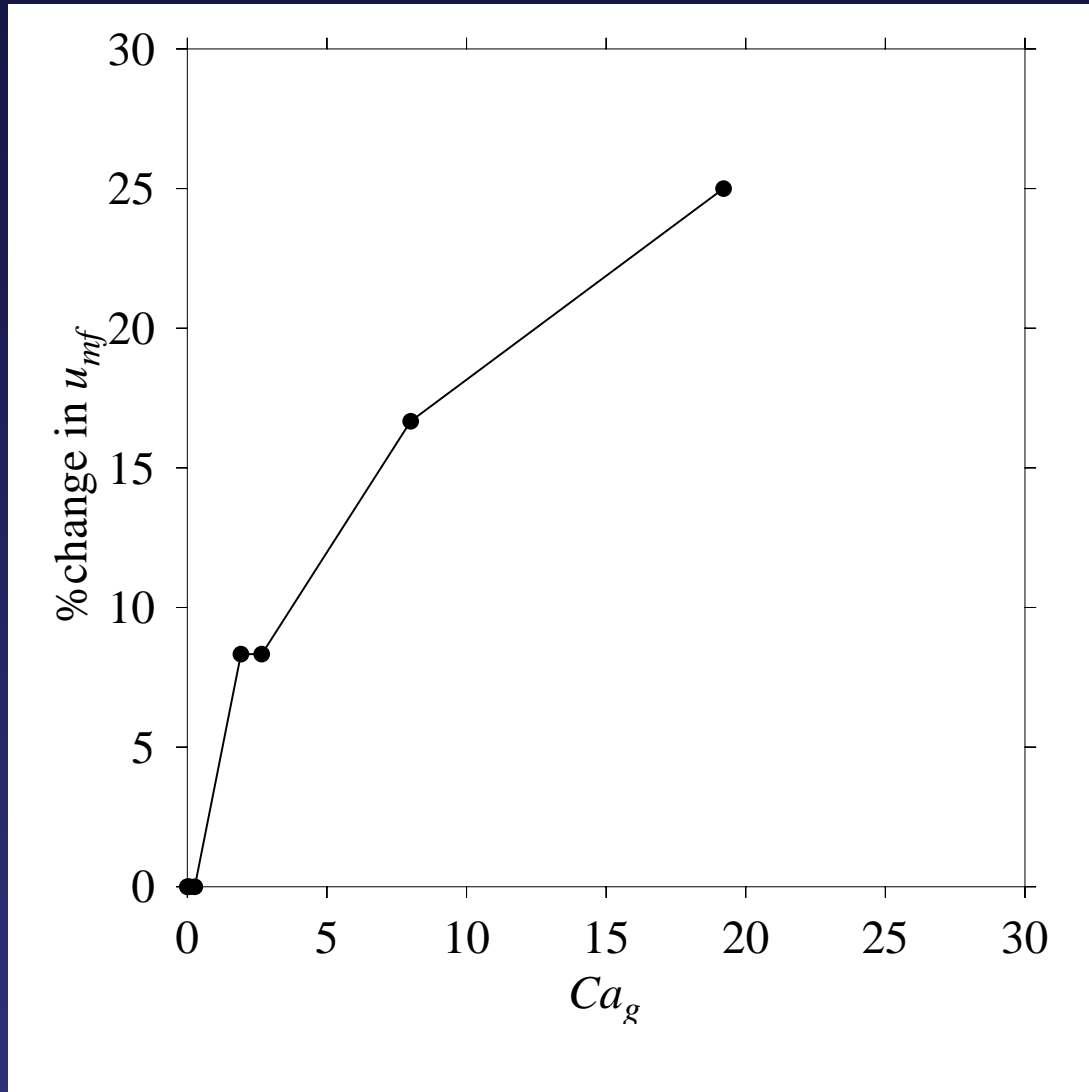
# The Capillary Number



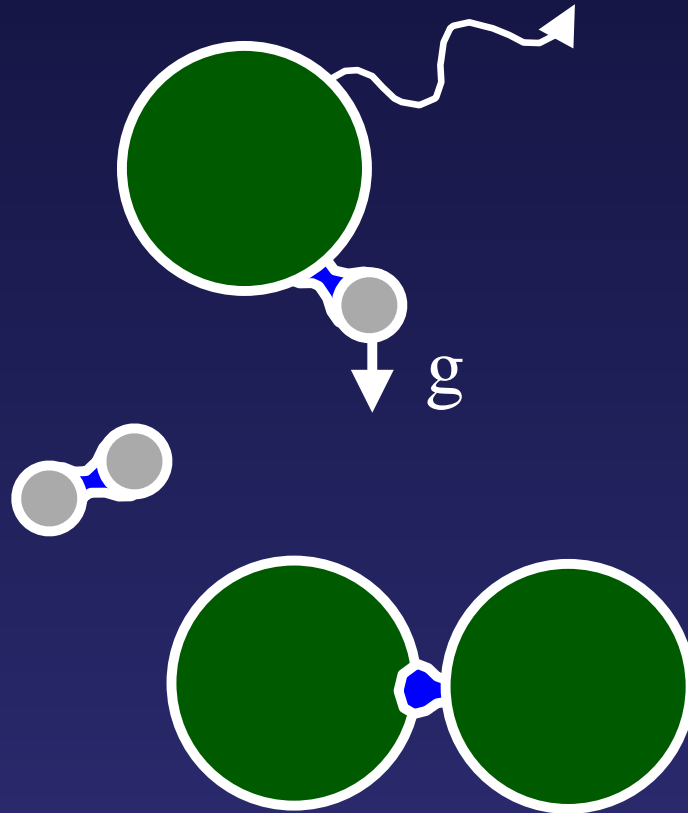
- $Ca_g = \frac{12\epsilon^3\gamma}{(1-\epsilon)(u-v_p)} \left[ \frac{1}{300(1-\epsilon)\mu_g + 7\rho_g(u-v_p)R} \right]$
- Two curves collapse with  $Ca_g$



# The Capillary Number (cont.)



# Micro-mixing Model – Static Systems

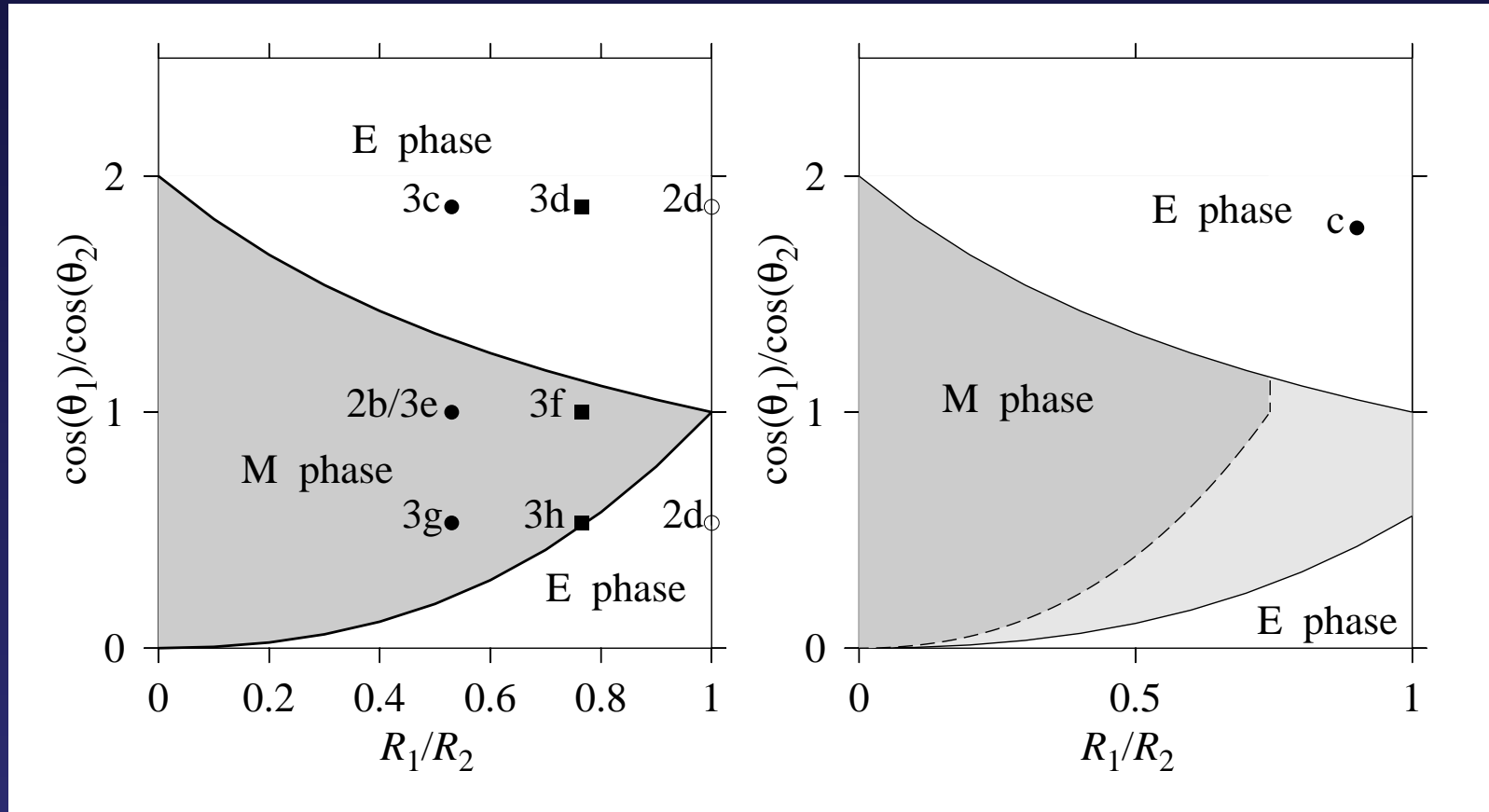


$$Bo_g = \frac{2\pi R_{eff}\gamma}{\frac{4}{3}\pi (\rho R^3)_{min} g}$$





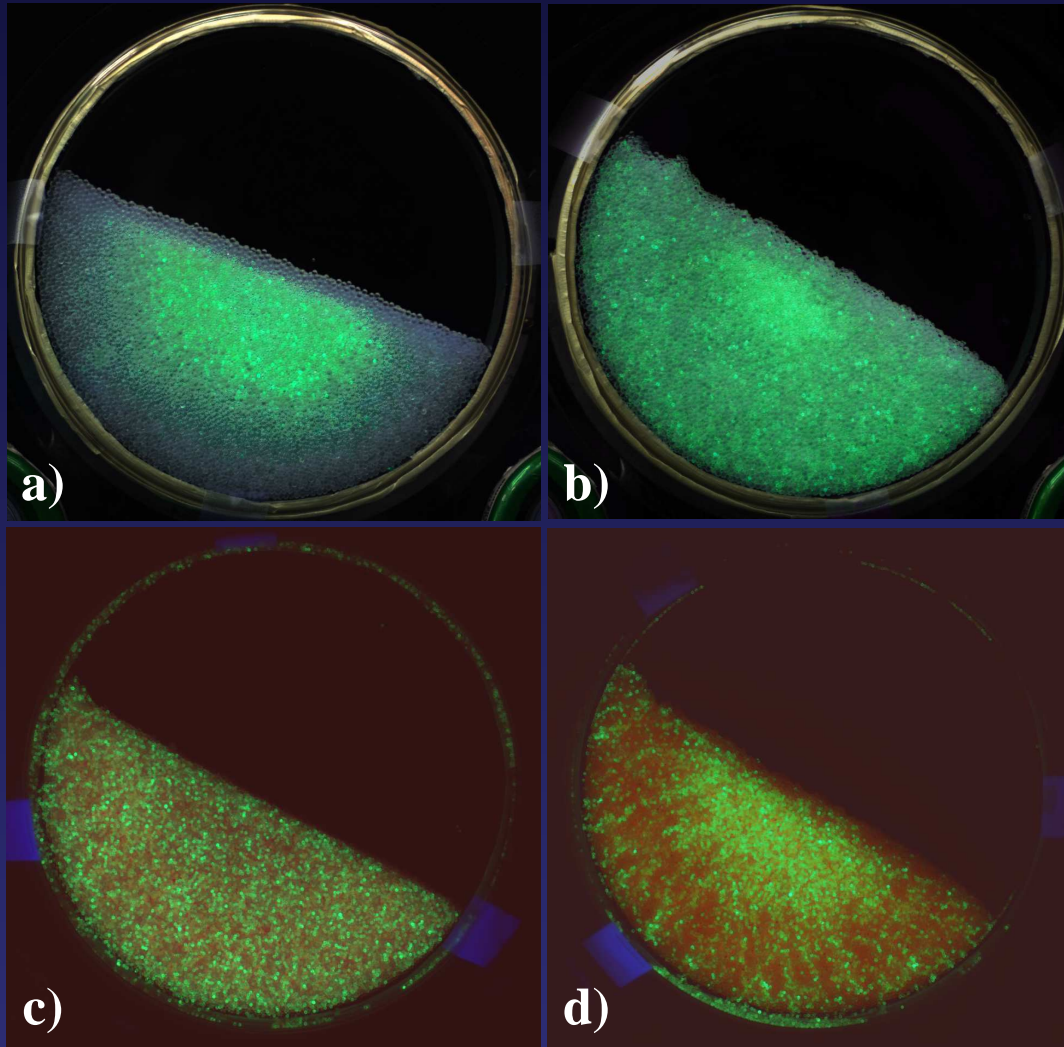
# A Phase Diagram



- Predict both “mixed” and segregated phases



# Manipulating the Phase Diagram

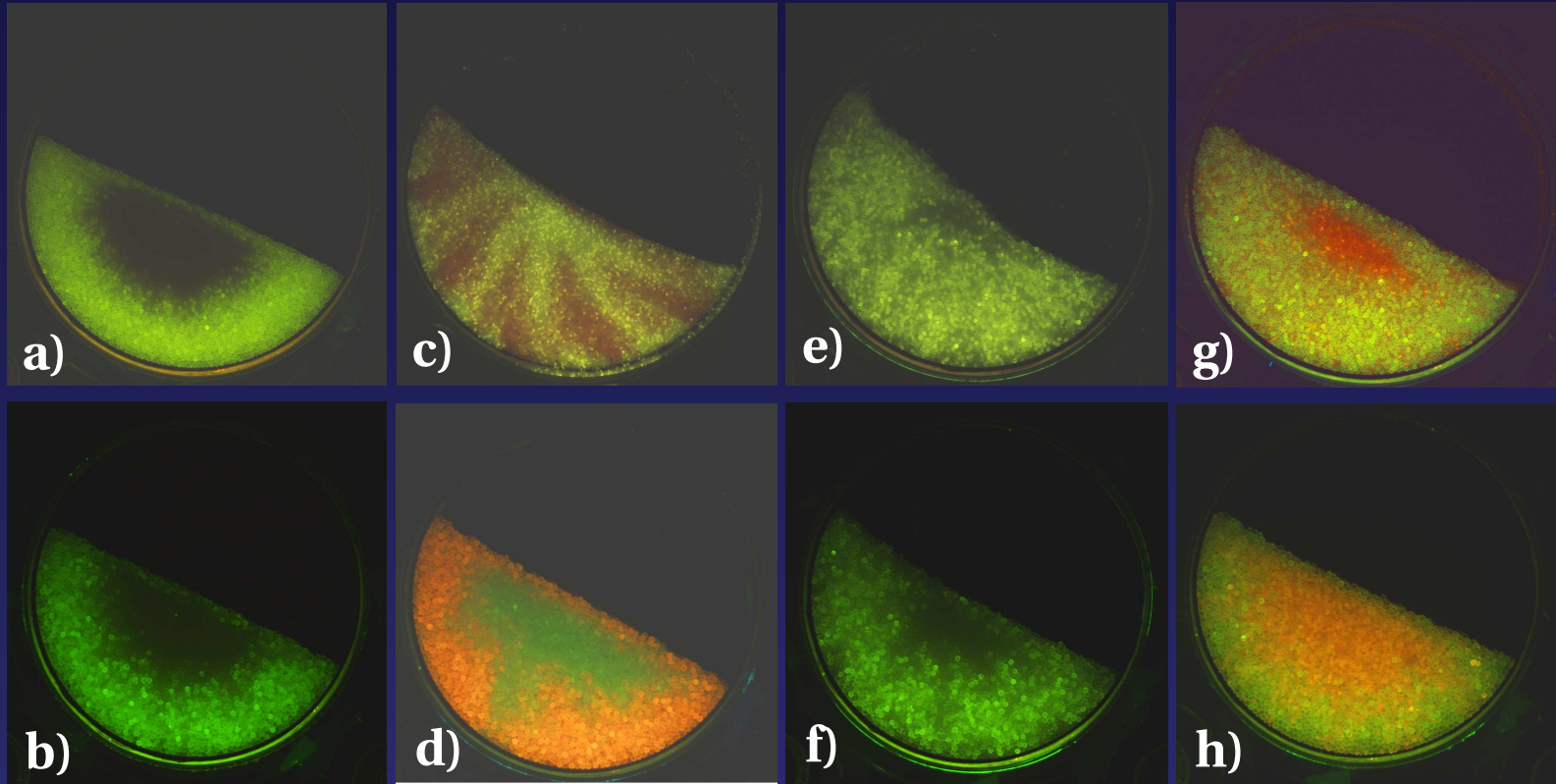


(PRL, 2003)

- Control mixing/segregation via surface properties!



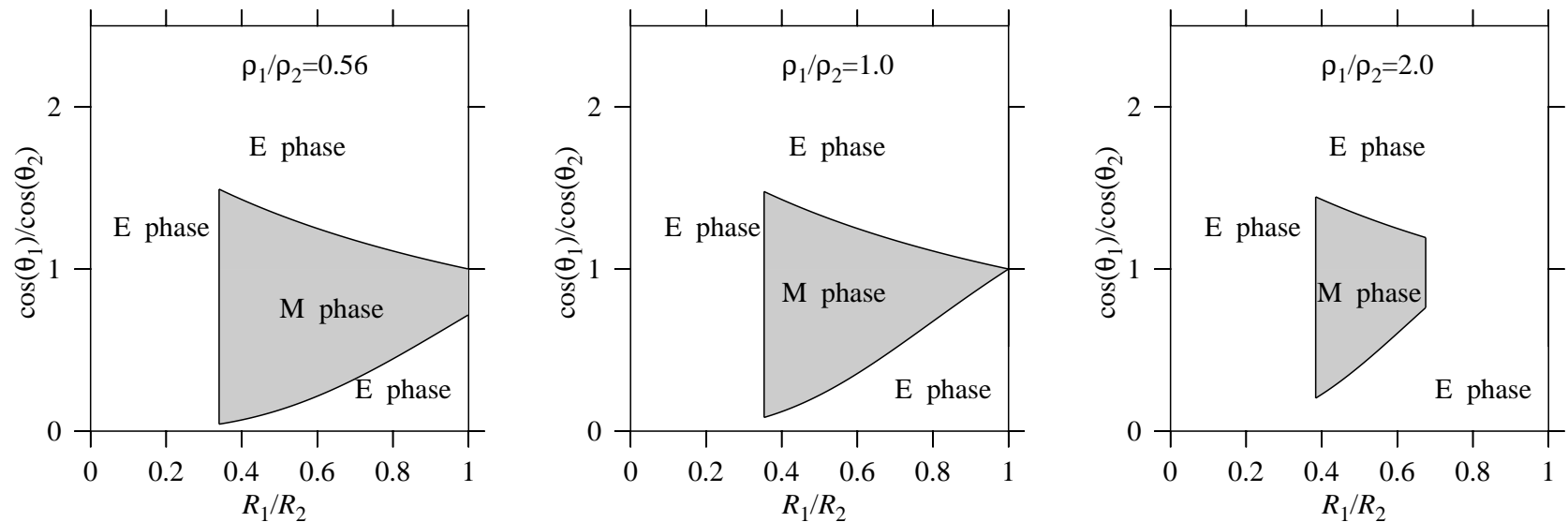
# Controlling Mixing/Segregation



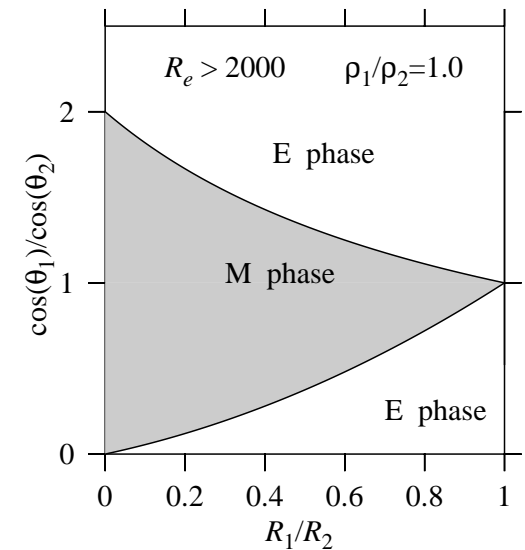
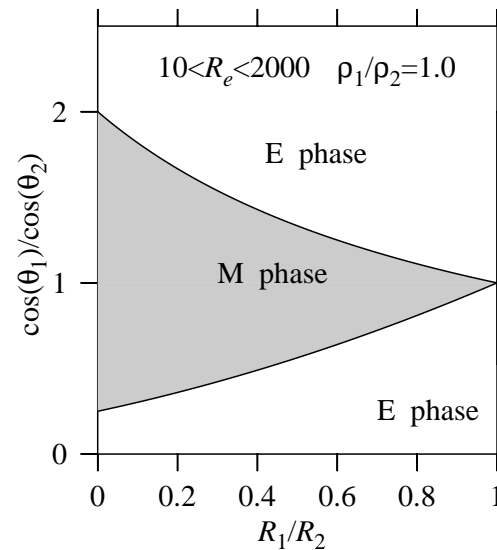
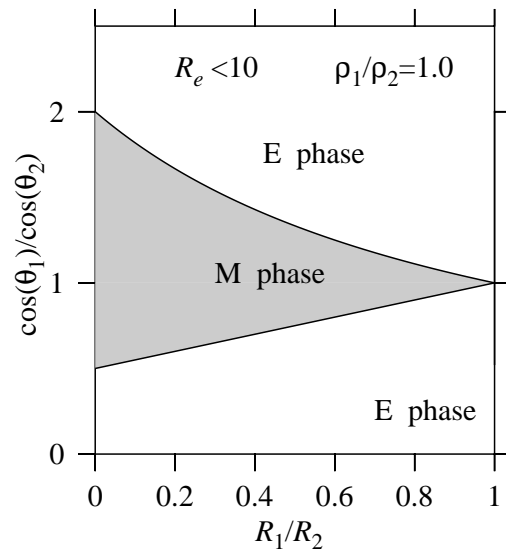
- Materials that would mix can be made to segregate and vice versa



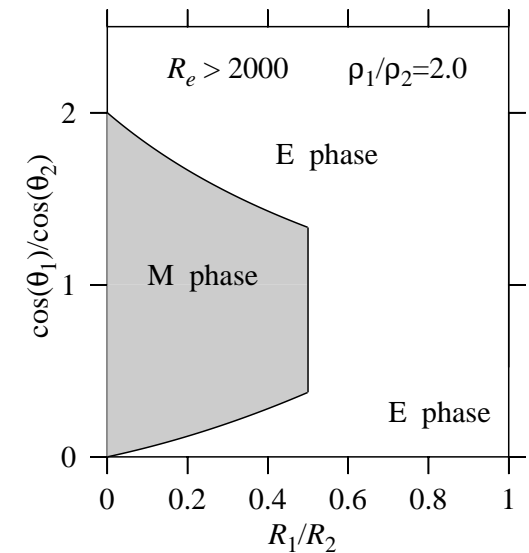
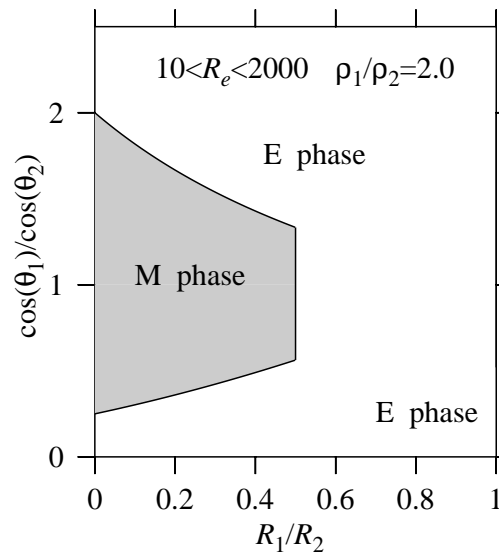
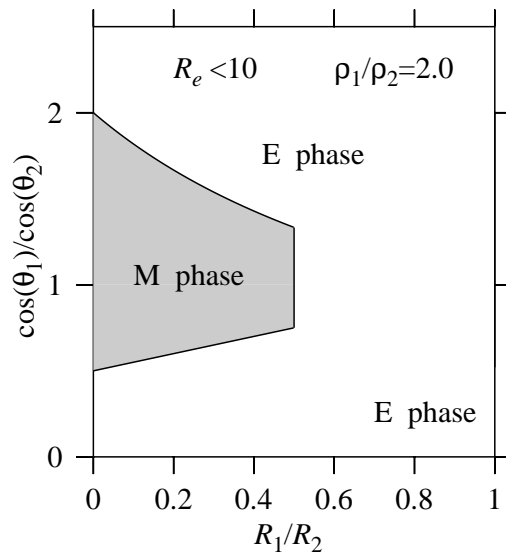
# Sheared Phases



# Gas-Solid Phases – Same Density



# Gas-Solid Phases – Diff. Density





# Outlook

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- Cohesion may be well characterized using discrete tools
- Controlling mixing/segregation works for psuedo-static systems
- Simulations/experiments in gas-solid and sheared systems are necessary



# Acknowledgments

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